

This article was downloaded by:

On: 27 January 2011

Access details: *Access Details: Free Access*

Publisher *Taylor & Francis*

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Nucleosides, Nucleotides and Nucleic Acids

Publication details, including instructions for authors and subscription information:

<http://www.informaworld.com/smpp/title~content=t713597286>

HPLC and LC-MS of Nucleosides

F. C. Alderweireldt^a; E. L. Esmans^a; P. Geboes^a

^a Laboratory for Organic Chemistry, University of Antwerp(RUCA), Antwerp, Belgium

To cite this Article Alderweireldt, F. C. , Esmans, E. L. and Geboes, P.(1985) 'HPLC and LC-MS of Nucleosides', *Nucleosides, Nucleotides and Nucleic Acids*, 4: 1, 135 — 137

To link to this Article: DOI: 10.1080/07328318508077837

URL: <http://dx.doi.org/10.1080/07328318508077837>

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: <http://www.informaworld.com/terms-and-conditions-of-access.pdf>

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

HPLC AND LC-MS OF NUCLEOSIDES.

★
F.C.Alderweireldt, E.L.Esmans, P.Geboes
University of Antwerp(RUCA)
Laboratory for Organic Chemistry
Groenenborgerlaan 171 B 2020 Antwerp Belgium

Summary.

An LC-MS system for the analysis and unambiguous identification of nucleosides is described using a microbore column and volatile buffer systems. The detection of pseudouridine in normal urine was used as a test for diagnostic applications.

For many years the area of nucleic acid research was hindered by the lack of good analytical techniques. Since nucleosides and nucleotides are heat-labile and low-volatile, gas chromatography cannot be used without derivatisation¹.

Development of high pressure liquid chromatography(HPLC) has proven to be very valuable for such separations. Originally low pressure ion exchange chromatography was used, but subsequently HPLC has proven to be particularly suitable using isocratic conditions as well as gradient elution^{1,2}.

However, in order for a chromatographic technique to be useful, peaks in biological samples must be identified unambiguously, for example by combining several methods such as retention times, use of standards, isotopic labelling, enzymic peak shift procedures or by using one single identification technique: mass spectrometry. Soft ionization techniques such as field desorption-, desorption/chemical ionization-, fast atom bombardment- and liquid chromatography-mass spectrometry(LC-MS) have proven to be most useful in the structure elucidation of such compounds. Amongst these mass spectrometric techniques LC-MS occupies a special place since it can be used for direct on-line analysis of mixtures. Experiments described in this area include nucleoside and nucleotide analysis via thermospray, moving belt and direct liquid introduction(DLI) interfacing. The thermospray interface gives good results not only

in the nucleoside but also in the nucleotide field^{3,4}. LC-MS using the moving belt interface seems to reveal molecular weight information if spectra are taken under chemical ionization conditions using ammonia as the reagent gas⁵.

DLI-interfacing of a mass spectrometer with a liquid chromatograph imposes specific restrictions upon the operating conditions of both HPLC and mass spectrometric part of the combined LC-MS system. This means that special conditions, using volatile salts such as ammonium acetate, ammonium formate etc. had to be established^{6,7}. Furthermore, due to cluster ion formation in the ion source, effluents must be chosen as a function of the molecular weight of the molecules to be determined. Since for most nucleosides a mass range from $m/z=110$ to $m/z=350$ has to be covered, the choice of organic solvents in HPLC is restricted towards methanol and acetonitrile. Even then, it is impossible to detect ions below $m/z=130$, unless a small oven is mounted between the DLI-probe and the ion source. Such a desolvation chamber was built in our workshop after the description given by Dedieu et al.⁸ but with slight modifications of the heating system. The impact of the insertion of this chamber into the DLI/LC-MS system was studied in combination with a μ -Bondapak 10RP18 column (30 cm \times 3.9 mm) using standard mixtures (30 μ g) of (deoxy)nucleosides at a flow-rate of 1 cc/min. Mass spectra could now be recorded from $m/z=110$ or $m/z=115$ depending upon the % of methanol in the effluent. As a result structurally important fragmentations such as $[BH + H]^+$, $[B + 30]^+$, $[B + 44]^+$ and $[B + 44 - 16]^+$ (deoxyribonucleosides) could now be observed.

Some kinds of diseases (gout or cancer) may give rise to "higher" levels of nucleosides in i.e. blood or urine. In order to obtain diagnostically interesting analysis of such urines, the detection limit of the system had to be improved. This was done by using a 50 cm \times 1 mm (Chrompack) 10 RP18 column operated at a flow-rate of 70 μ l/min. This flow-rate implies a split factor of 1/7 instead of 1/100 if flow-rates of 1 cc/min were used. Under these conditions some severe difficulties had to be overcome due to corrosion of the nickel diaphragm in the interface by the effluent. Once these difficulties were solved, good separations and spectra were obtained for standard mixtures (1.5 μ g) of cytidine, uridine, 2'-deoxycytidine e.g., using 97% 0.01 M NH_4OOCCH_3 /3% CH_3OH as the effluent.

Finally 20 cc of morning urine was cleaned-up² using Affigel 601 (Biorad) and analyzed on the microbore DLI/LC-MS system using the same chromatographic conditions as described above. Pseudouridine could be identified by monitoring $m/z=245$ (MH^+) and by comparison with the mass spectrum of a pseudouridine standard. The possibilities of this technique will now be studied for the analysis of pathological urine samples.

REFERENCES.

- 1.K.H.Schramm,J.A.Mc Closkey in "GLC and HPLC determination of therapeutic agents",part 3,Ed.K.Tsuji,p.1149,Marcel Dekker,New-York(1979).
2. C.W. Gehrke,K.C.Kuo,G.E.Davis,R.D.Suits,T.P.Waalkes,E.Borek, Journ.Chromatogr.,150,455(1978).
- 3.C.R.Blakley,M.J.Mc Adams,M.L.Vestal,Journ.Chromatogr.,158,261(1978).
- 4.C.G.Edmonds,H.Pang,J.A.Mc Closkey,presented at the "30th Annual Conference of Mass Spectrometry and Allied Topics" Honolulu,Hawaii(1982), abstracts p.606.
- 5.D.E.Games,P.Hirter,W.Kunh,E.Lewis,N.C.A.Weerasinghe,S.A.Westwood, Journ.Chromatogr.,203,131(1981).
- 6.E.L.Esmans,Y.Luyten,F.C.Alderweireldt,Biomed.Mass Spectrom.,10, 347(1983).
- 7.E.L.Esmans,Y.Luyten,F.C.Alderweireldt,P.Krien,G.Devant,presented at the "30th Annual Conference on Mass Spectrometry and Allied Topics" Honolulu,Hawaii (1982),abstracts p.608.
- 8.M.Ddedieu,C;Juin,P.J.Arpino,G.Guiochon,Anal.Chem.,54,2372(1982).

ACKNOWLEDGEMENTS.

We thank the NFWO for financial support.This work is also supported by NATO grant 424/84.